### Identification

NaOH

M = 40,00 g/mol CAS [1310-73-2]

EC number: 215-185-5 Taric code: 2815 12 00

## **Applications**

analytical chemistry, titrant in volumetric analysis.

#### **Specifications**

uncertainty ± 0,001

1 ml = 0,0400 g NaOH

This solution was analysed using a certified reference material (potassium hydrogen phthalate).

The certified reference material is ISO 17034 accredited, measured according to ISO/IEC 17025 and traceable to the International System of Units by means of a Standard Reference Material from NIST: SRM® 84 (Potassium Hydrogen Phthalate).

#### Physical data

- Density: 1,04 g/cm3
- pH(20 °C) ~ 13,7

## Safety - GHS

Signal Word: Danger

**Hazard Statements:** 

H314: Causes severe skin burns and eye damage.

# E.S.

#### **Precautionary Statements:**

P260: Do not breathe dust / fume / gas / mist / vapours / spray.

P303+P361+P353: IF ON SKIN (or hair): Remove / Take off immediately all contaminated clothing. Rinse skin with water / shower.

P305+P351+P338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P321: Specific treatment (see on this label).

P405: Store locked up.

P501a: Dispose of contents / container in accordance with local / regional / national / international regulations.

### **Toxicological data**

- MAK: 2 mg/m³ (pure substance)
- WGK: 1
- Poison class CH (Swiss): 3

## Transport/storage

- ADR: 8 C5 II UN 1824 SODIUM HYDROXIDE SOLUTION
- IMDG: 8 II UN 1824 SODIUM HYDROXIDE SOLUTION
- IATA/ICAO: 8 II UN 1824 SODIUM HYDROXIDE SOLUTION
- PAX: 809
- CAO: 813
- Store between 15°C and 25°C